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WordWise. Solution. A mixture that results when substances dissolve to form a homogeneous mixture.

Compound. A substance that can be

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broken down into two or more simpler substances. Physical Change. A change which the composition of matter stays the same.

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Properties of Matter. Section 2.3  
Chemical Properties. (pages 54–58). This  
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and describes clues that may show.

Section 2.3 Chemical Properties (pages 54–58) This section discusses chemical properties and describes clues that may show that a chemical change has taken place.

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□\*\*Every sample of a given substance has the same properties because a substance has a fixed, uniform composition. □Substances are classified into two categories: elements and compounds. □An element is a substance that cannot be broken down into simpler substances. □An atom is the smallest particle of an element.

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A property that causes subatomic particles such as protons and electrons to attract or repel other matter 2. The attraction or repulsion between electrically charged objects 3. Charge transfer without contact between

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materials 4. Law that total charge in an isolated system is constant 5. A continuous flow of electric charge 6.

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Chapter 2 Properties of Matter Section 2.3 Chemical Properties (pages 54–58)  
This section discusses chemical

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properties and describes clues that may show that a chemical change has taken place. Reading Strategy (page 54)  
Relating Text and Visuals As you read, complete the table by finding examples of the clues for recognizing chemical changes in

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## **Section 2.1 Classifying Matter**

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- Physical change do not change the identity of the substance - the matter stays the same.
- Chemical change happens when 2 or more substances are changed into NEW substances with DIFFERENT properties.
- Physical

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property of matter can be observed without changing the identity of the matter.

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### **Chapter 2**

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